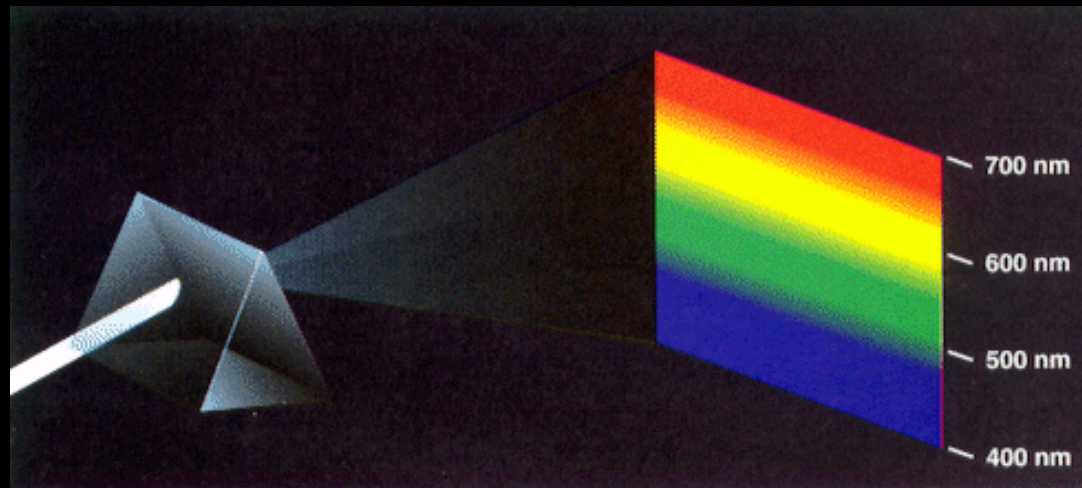


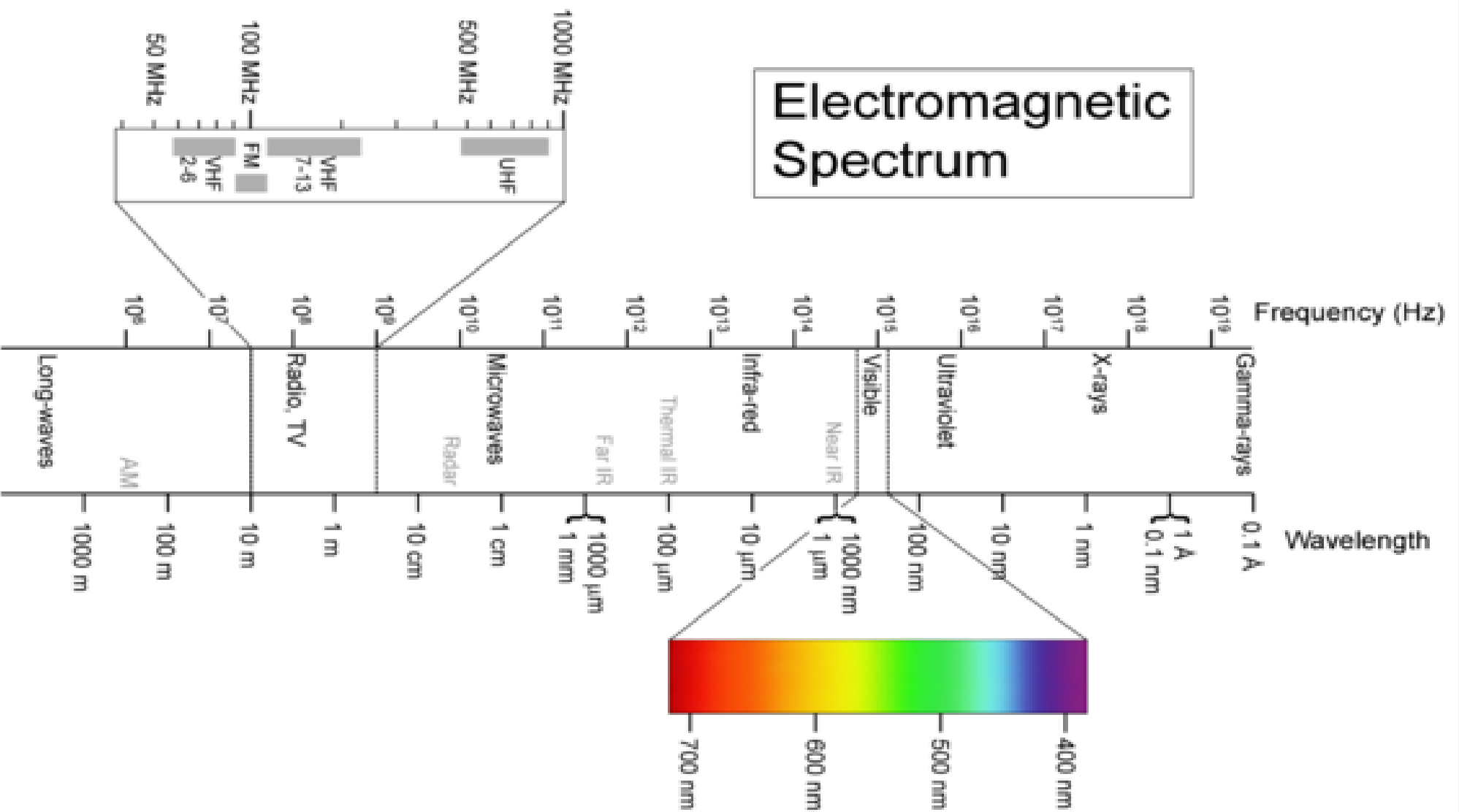
AMATEUR SPECTROSCOPY



Percy Jacobs
Pretoria ASSA Center
2016

Spectroscopy is the study of the different wavelengths/frequencies of light we see from an object. It is a measure of the quantity of each colour of light (or more specifically, the amount of each wavelength of light). It is a powerful tool in astronomy. In fact, most of what we know in astronomy is a result of spectroscopy: it can reveal the temperature, velocity and composition of an object as well as be used to infer mass, distance and many other pieces of information. Spectroscopy is done at all wavelengths of the electromagnetic spectrum, from radio waves to gamma rays; but here we will focus on optical light.

Electromagnetic Spectrum



wavelength



650 nm



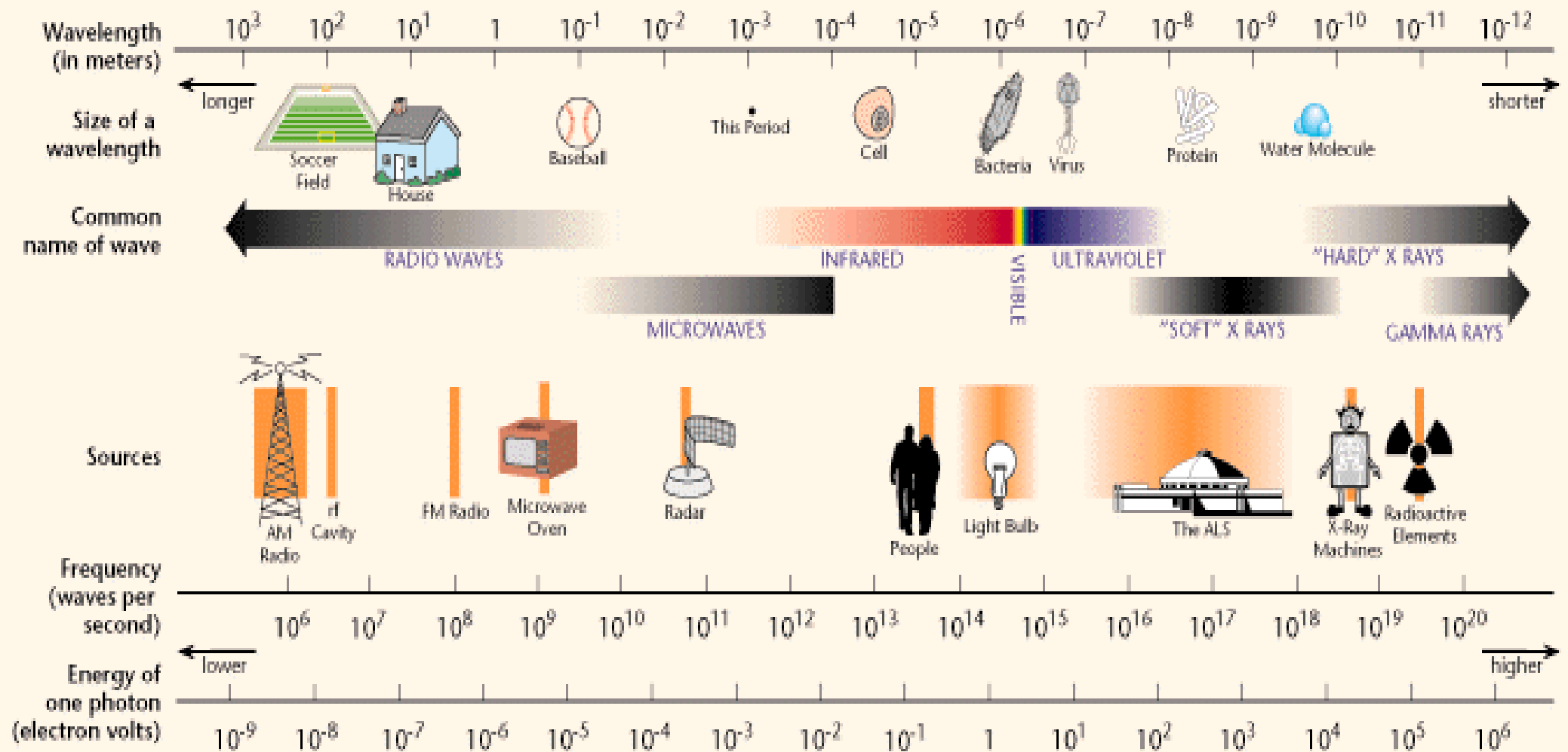
wavelength



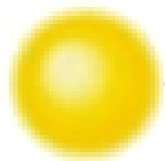
400 nm



THE ELECTROMAGNETIC SPECTRUM



light
source



A hot solid

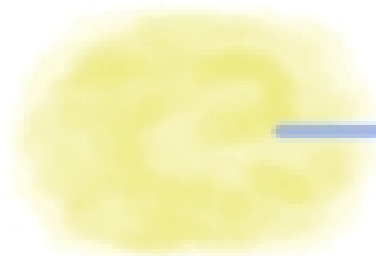
absorption

prism

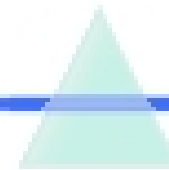
result



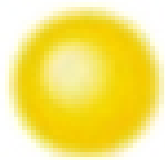
Continuous
spectrum



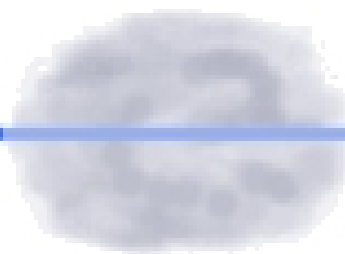
A hot gas
at high pressure



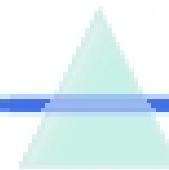
Emission line
spectrum



A hot solid



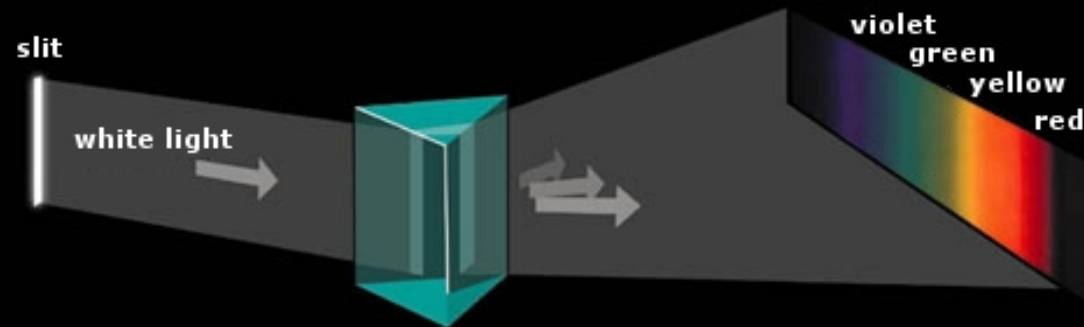
A cold gas
at a low pressure



Absorption line
spectrum

Equipment – Visual Spectroscopy

. Prism



. CD or DVD



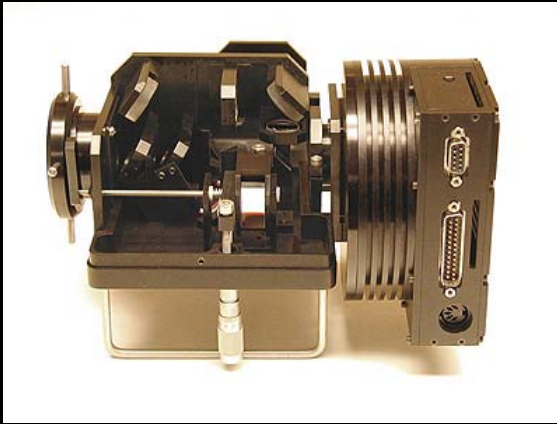
. Naked Eye



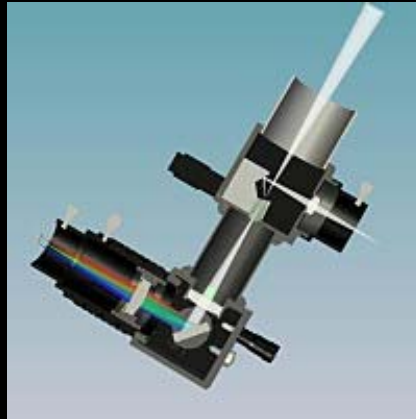
Equipment – Telescopic Spectrographs

More Expensive – R15,000 to R85,000

SBIG



Baader DADOS

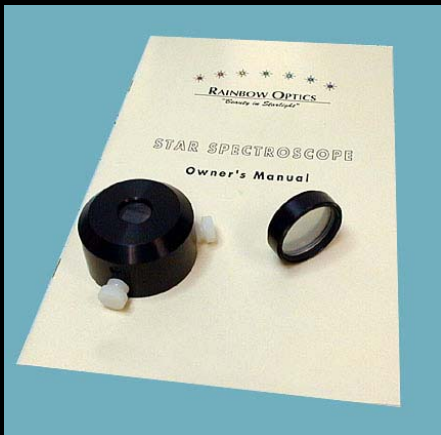


Lhires Lite



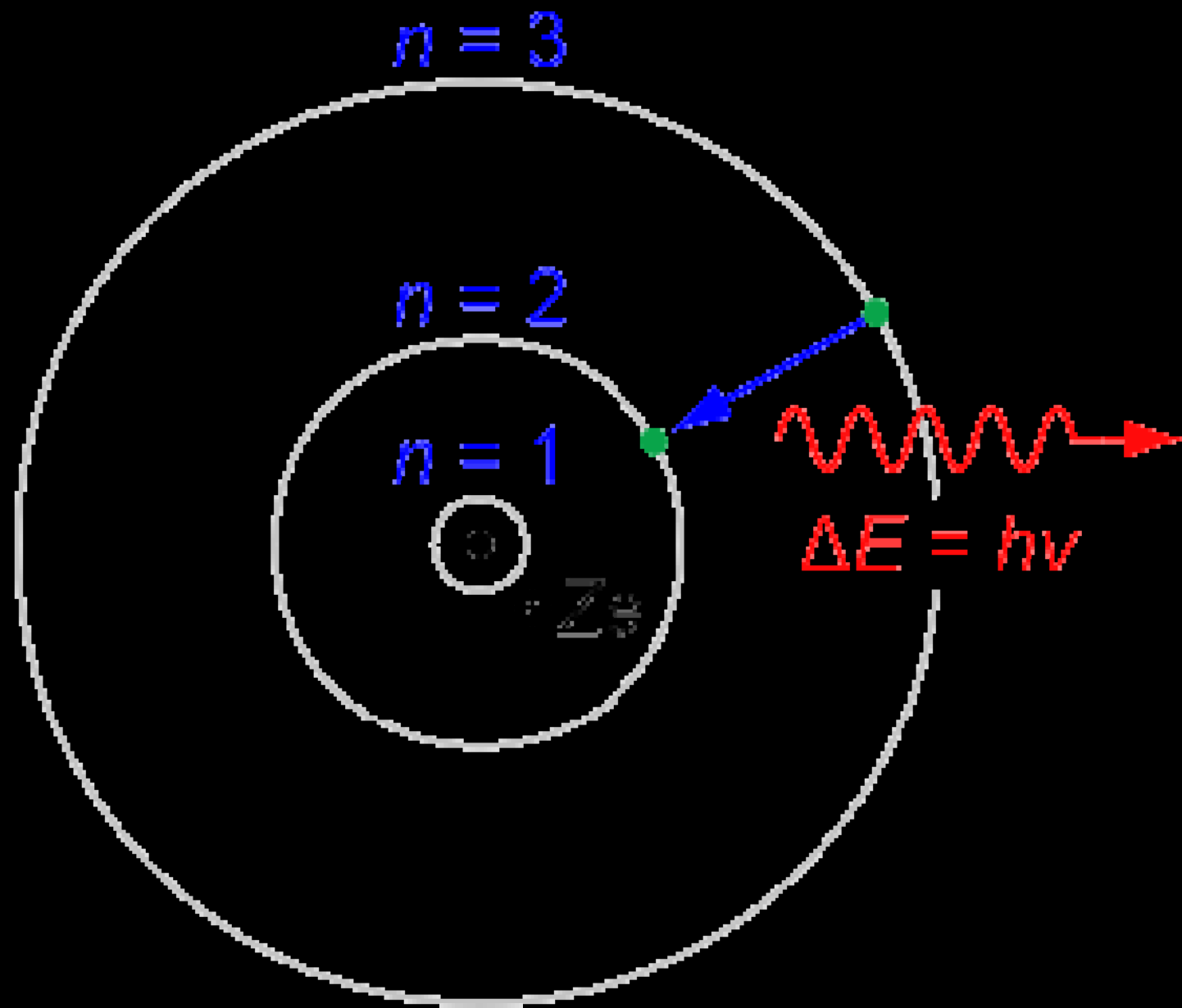
Less Expensive – R2,000 to R3,000

Rainbow Optics Star Spectroscopes



The Star Analyser 100 (SA-100 or 200)

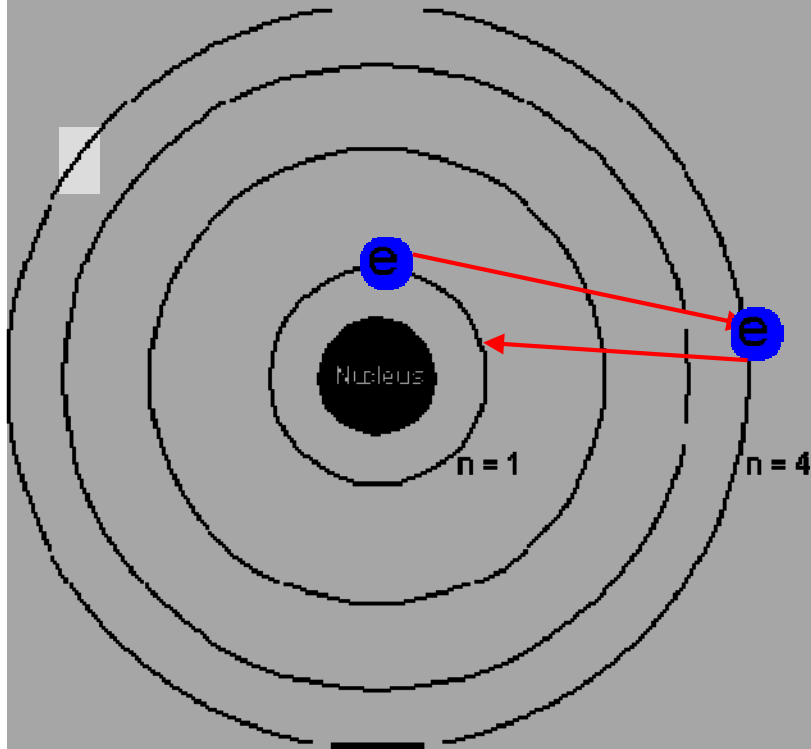




GROUND STATE

A hydrogen atom in its Ground State.

The electron occupies the lowest possible energy level which in the case of hydrogen is the Principal Quantum Level $n = 1$.



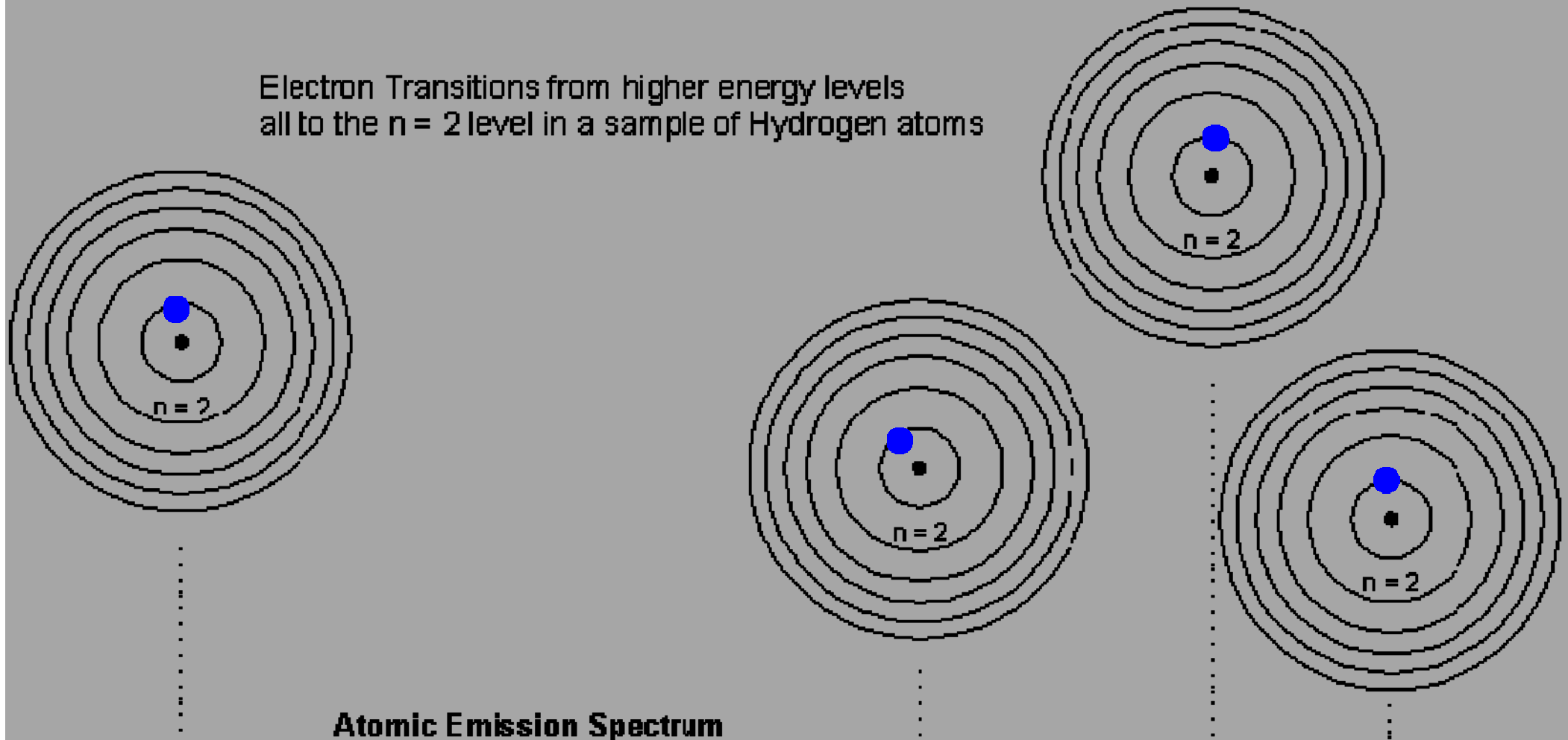
EXCITED STATE – absorbs energy – a specific amount of photon energy

A hydrogen atom in an Excited State.

The electron occupies one of the higher energy levels further from the nucleus of the atom.

Ground State

Electron Transitions from higher energy levels
all to the $n = 2$ level in a sample of Hydrogen atoms



Atomic Emission Spectrum

656.3 nm

Balmer Series - Visible Region

486.1 nm

434.0 nm

410.2 nm

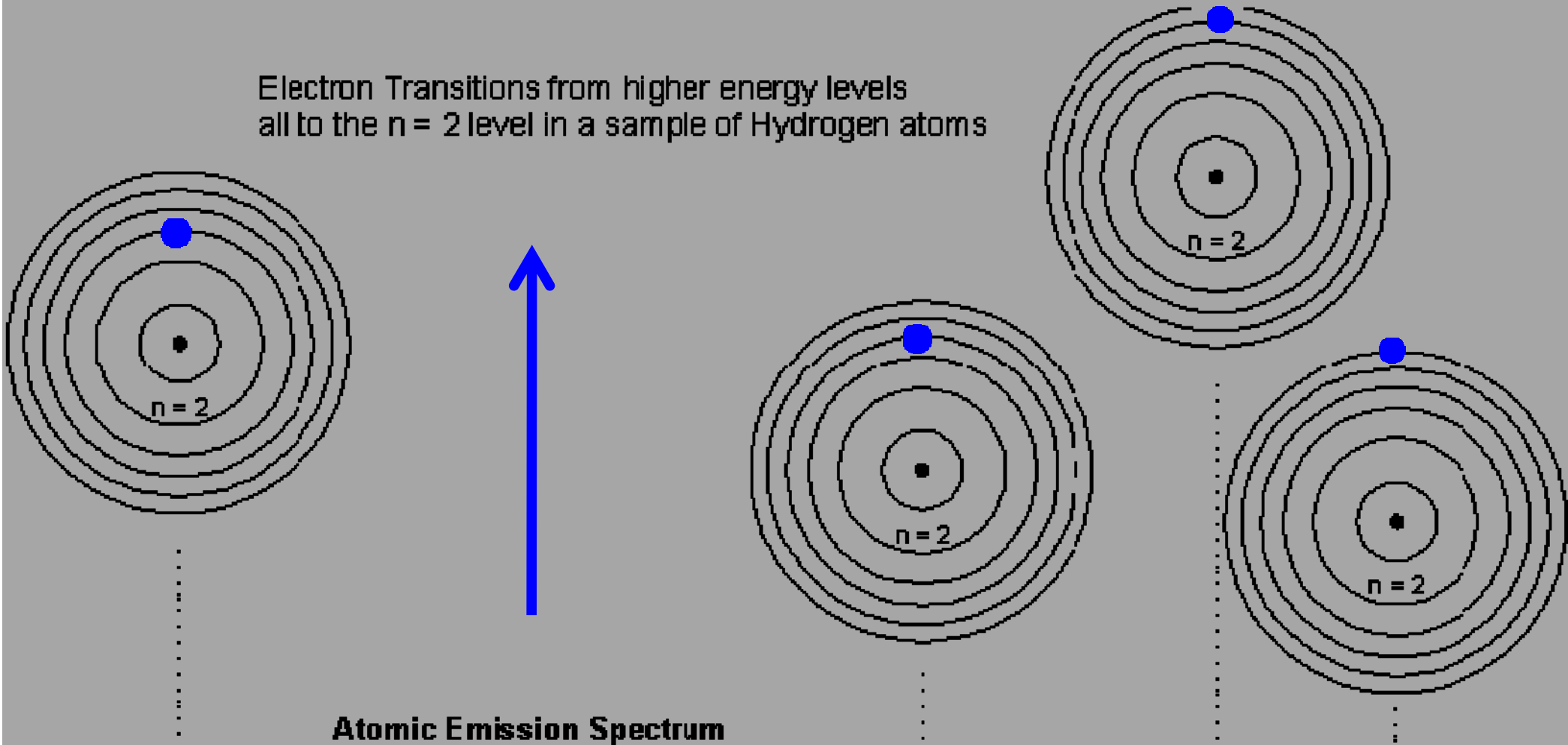
← Increasing Wavelength

Increasing Frequency →

Exited State (absorb energy)

- H α 6563 – n2 / n3
- H β 4861 – n2 / n4
- H γ 4340 – n2 / n5
- H δ 4101 – n2 / n6

Electron Transitions from higher energy levels
all to the n = 2 level in a sample of Hydrogen atoms



Atomic Emission Spectrum

Balmer Series (emits energy)

- H α 6563 – n3 / n2
- H β 4861 – n4 / n2
- H γ 4340 – n5 / n2
- H δ 4101 – n6 / n2

Electron Transitions from higher energy levels
all to the n = 2 level in a sample of Hydrogen atoms

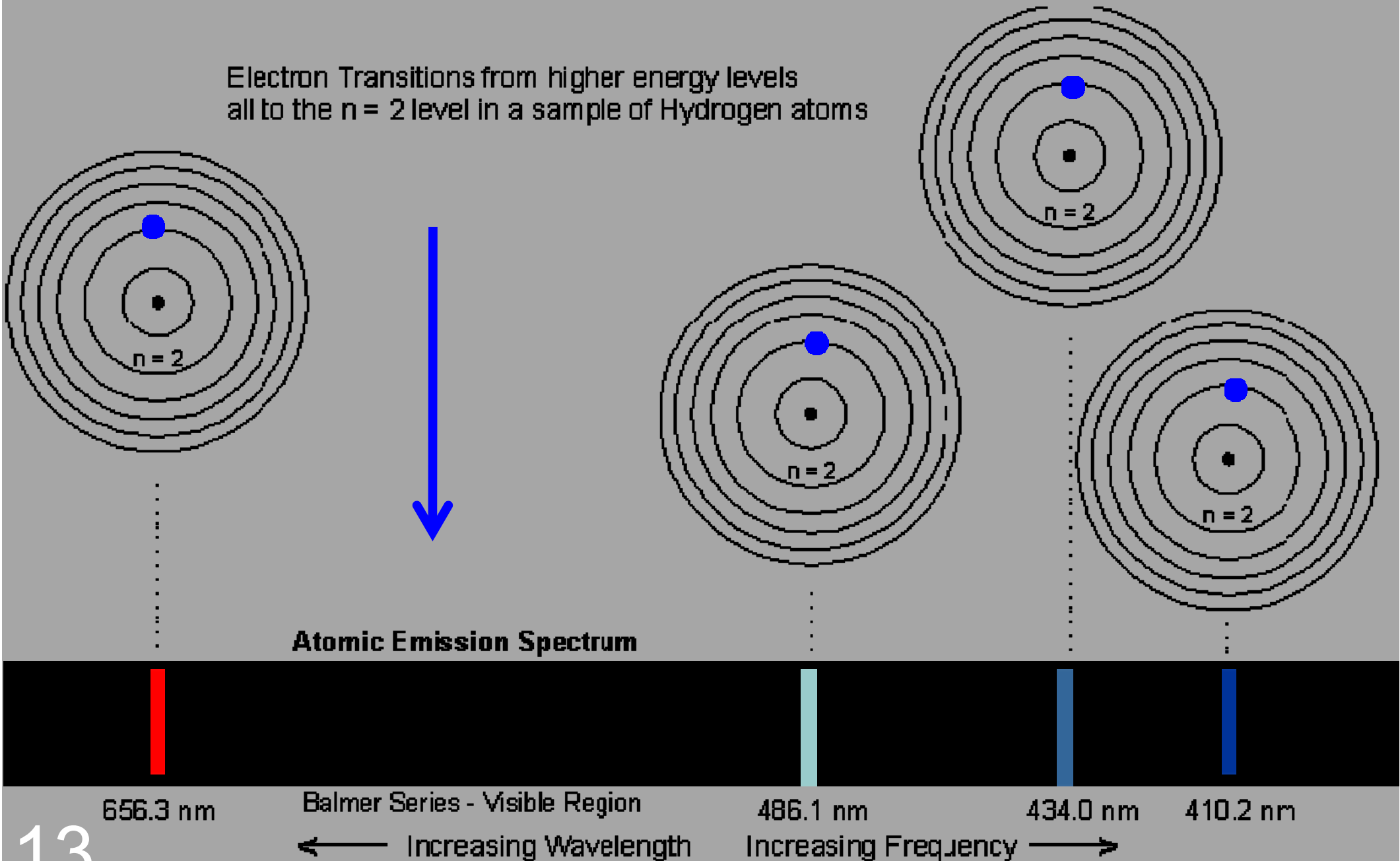
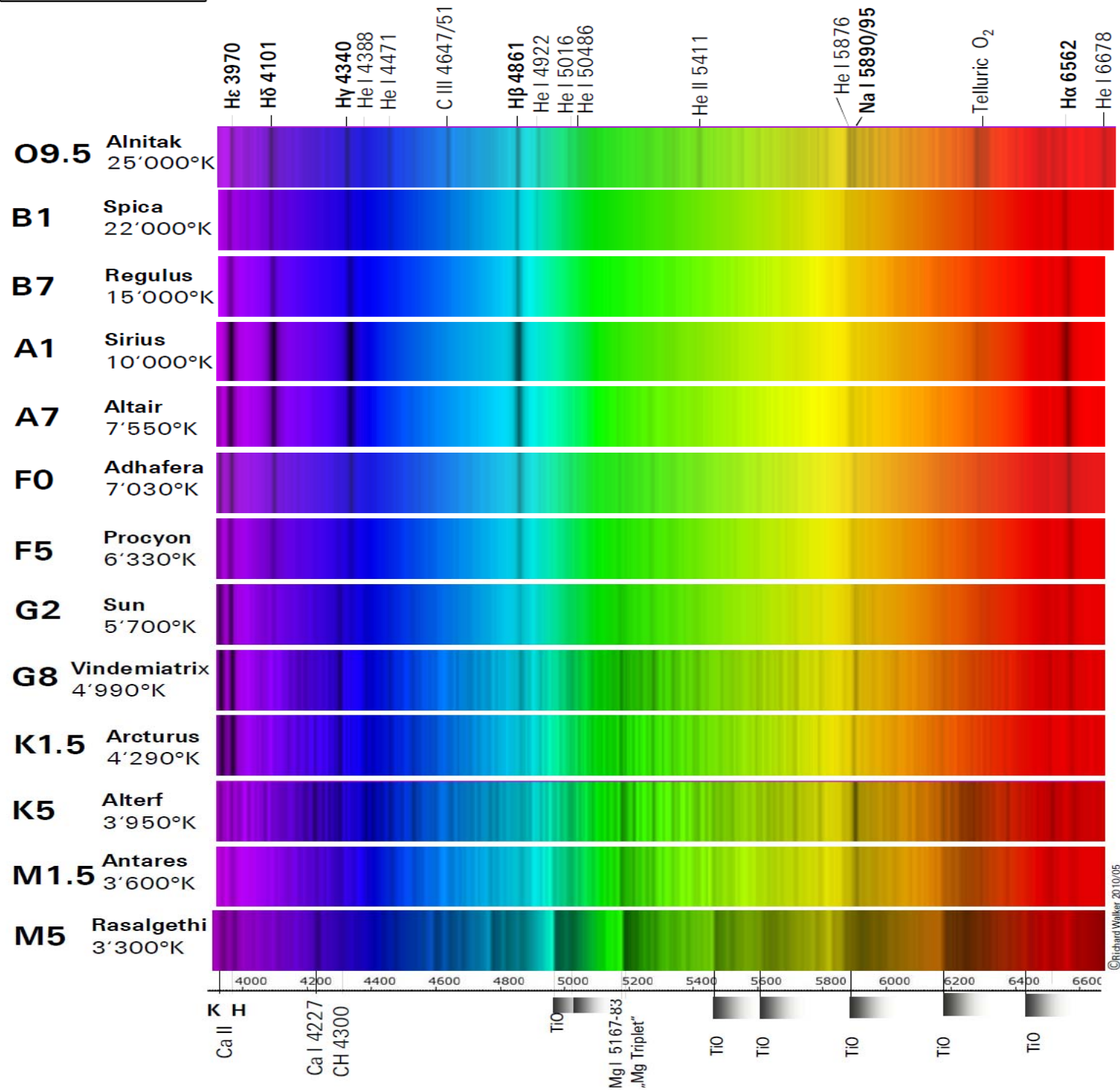


TABLE 01

Overview on the spectral classes

©Richard Walker 2010/05

Class O - Lamda Orionis - Blue > 30,000K



- Class O stars are very hot and very luminous, being bluish in color. These are the rarest of all main sequence stars.
- Helium is very difficult to ionize, so spectral lines by ionized Helium (He II) only appear in these hot stars, prominent ionized Si IV, OIII, NIII, CIII,

Class B - Rigel, Spica - Blue-White 9500K - 30,000K



- Class B stars are extremely luminous and blue
 - They are hot enough to energize their Helium, but are not hot enough to ionize it. Thus *B* stars have HeI lines (neutral He) but do not have HeII lines
 - Ionized metal lines include MgII, SiII
- “**B & O stars**” – stars with temp’s > 10,000 K, most of the Hydrogen gas in the star’s atmosphere will be ionized. Since an ionized Hydrogen atom has no electron it cannot produce any spectral lines, thus the Hydrogen lines are weak

Class A - Sirius, Vega, Altair - White 7000K - 9500K



- Class A stars are amongst the more common naked eye stars, and are white or bluish-white.
- They have strong hydrogen lines and also lines of ionized metals FeII, MgII, SiII, CaII
- They do not have any Helium lines at all

Class F - Canopus, Procyon - Yellow-White 6000K - 7000K



- Class F stars have strengthening CaII lines + neutral metals FeI, CrI
- These stars are within the right range of temperatures to energize their Hydrogen gas without ionizing it. Thus the Hydrogen “Balmer” lines are very strong in these stars.

Class G - The Sun, Capella - Yellow 5200K – 6000K



- Most notable are the lines of CaII.
- They have even weaker hydrogen lines than **Class F**, but along with the ionized metals, they have neutral metals.

Class K - Arcturus, Aldebaran - Orange 3900K – 5200K



- Class K are orangish stars which are slightly cooler than our Sun.
- Some K stars are giants & supergiants.
- They have extremely weak hydrogen lines, if they are present at all, and mostly neutral metals MnI, FeI, SiI.
- By late K, molecular bands of TiO become present.

Class M - Antares, Betelgeuse - Red < 3900K



- Class M are most common.
- The spectrum of an **M star shows lines belonging to molecules** and all neutral metals but hydrogen lines are usually absent. Titanium oxide can be strong in M stars. Vanadium oxide bands are also seen in late M stars.

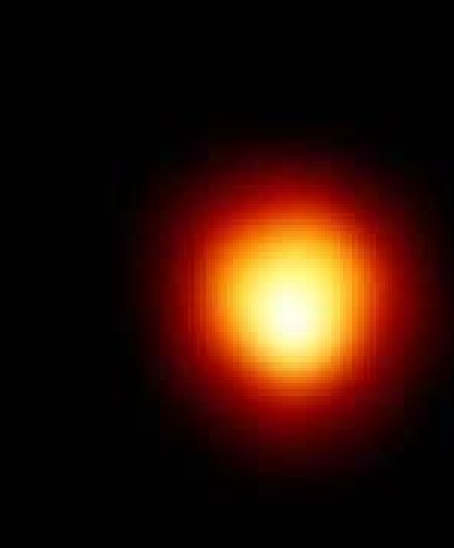
Class W or WR - Wolf-Rayet stars – Gamma Velorum

- Unusual since they have mostly helium in their atmospheres instead of hydrogen. They are thought to be dying supergiants with their hydrogen layer blown away by hot stellar winds caused by their high temperatures, thereby directly exposing their hot helium shell. They show spectral lines of carbon, nitrogen, or oxygen emission in their spectra (and outer layers).
- These lower temp stars, the Hydrogen gas isn't as easily excited, thus the Balmer lines aren't as strong

Class W: Wolf-Rayet



Class G & M



Class B



Common Spectral Lines (given in Å):

Hydrogen (the “Balmer Series”):

H α 6563
H β 4861
H γ 4340
H δ 4101
H ϵ 3970
H8 3889
H9 3835
H10 3798
H11 3771
H12 3750

Molecular Bands:

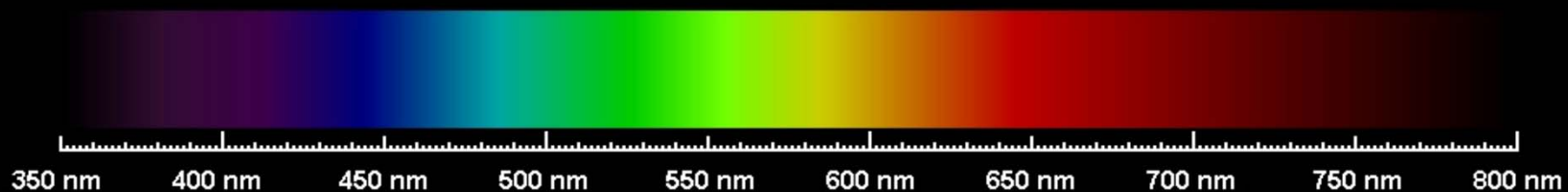
CH “G band” 4300
CN 3880, 4217, 7699
C2 “Swan” 4380, 4738, 5165, 5635, 6122
C3 4065
MgH 4780
TiO 4584, 4625, 4670, 4760

Helium:

He I 4026, 4388, 4471, 4713, 5015, 5048, 5875, 6678,
He II 4339, 4542, 4686, 5412

Metals:

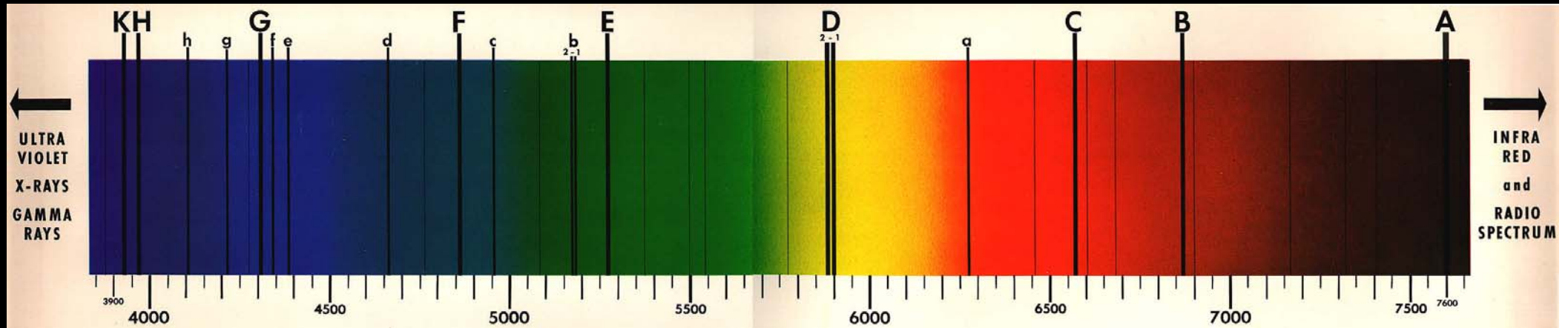
| | |
|---------------------------|------------------|
| C II 4267 | Si III 4552 |
| C III 4649, 5696 | Si IV 4089 |
| C IV 4658, 5805 | Ca I 4226 |
| N III 4097, 4634 | Ca II 3933, 3968 |
| N IV 4058, 7100 | Sc II 4246 |
| N V 4605 | Ti II 4300, 4444 |
| O V 5592 | Mn I 4032 |
| Na I 5890, 5896 | Fe I 4045, 4325 |
| Mg I 5167, 5173, 5183 | Fe II 4175, 4233 |
| Mg II 4481 | Sr II 4077, 4215 |
| Hg 4358, 5461, 5770, 5791 | |



Visible Continuous Spectrum 2

(Perceived Brightness Partially to Scale)

Fraunhofer lines – Sun Spectrum



| Lines | Due To | Wavelengths | Lines | Due To | Wavelengths |
|------------|----------------|-------------|-------|---------|-------------|
| A - (band) | O ₂ | 7594 - 7621 | F | H | 4861 |
| B - (band) | O ₂ | 6867 - 6884 | d | Fe | 4668 |
| C | H | 6563 | e | Fe | 4384 |
| a - (band) | O ₂ | 6276 - 6287 | f | H | 4340 |
| D - 1, 2 | Na | 5896 & 5890 | G | Fe & Ca | 4308 |
| E | Fe | 5270 | g | Ca | 4227 |
| b - 1, 2 | Mg | 5184 & 5173 | h | H | 4102 |
| c | Fe | 4958 | H | Ca | 3968 |
| | | | K | Ca | 3934 |

Hydrogen



Carbon



Helium



Nitrogen



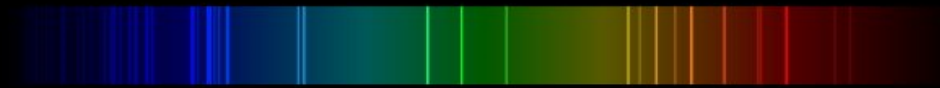
Lithium



Neon



Oxygen



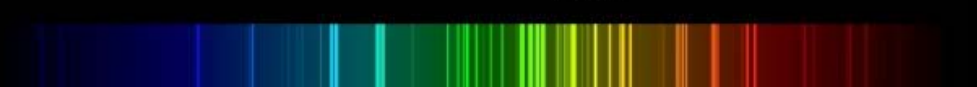
Magnesium



Calcium



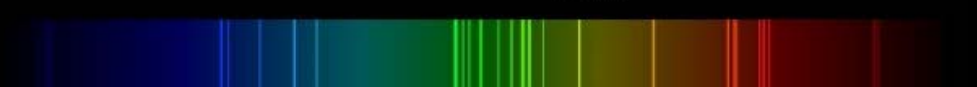
Silicon



Argon



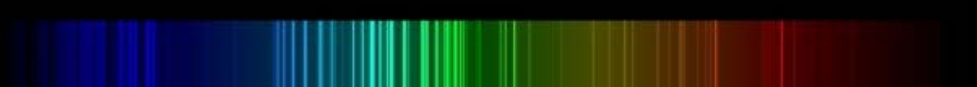
Sulfur



Sodium



Iron



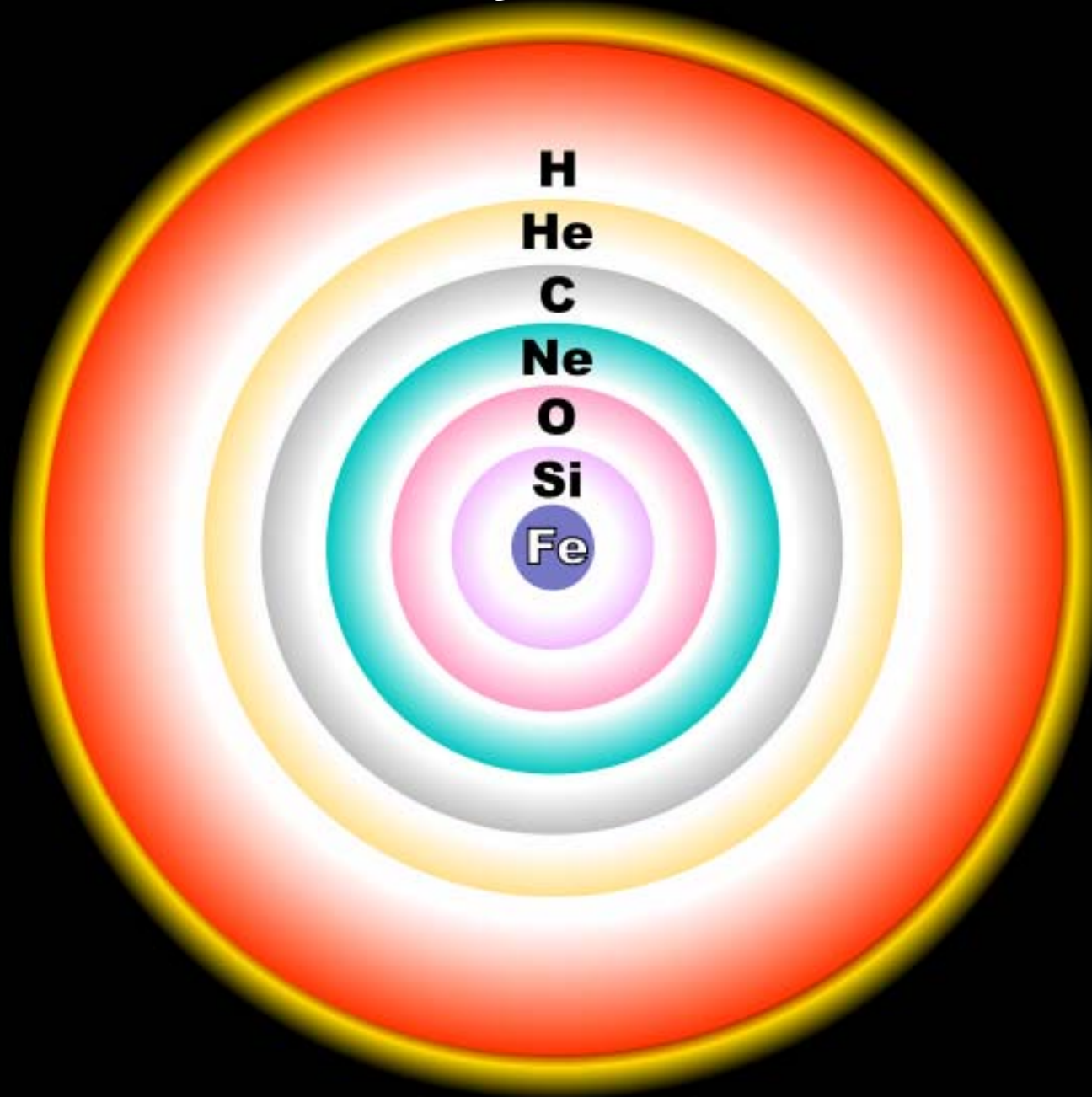
Krypton



Aluminum



"Onion" Layers of the Sun



Author's equipment:

80 mm refractor,
Rainbow Optics Transmission Grating (200l/mm),
Nikon DSLR,
An equatorial mounting fitted with a tracking unit.



This shows the various instruments on the equatorial mounting. Left is the 80 mm refractor with the DSLR camera and grating fitted into the refractor tube where the eyepiece is currently (on the left), with the finder scope on the top. To the right is a 400 mm telephoto lens fitted with a ZWO130 CCD camera, used as a guide telescope.



4 Processing the Image to get a spectrum

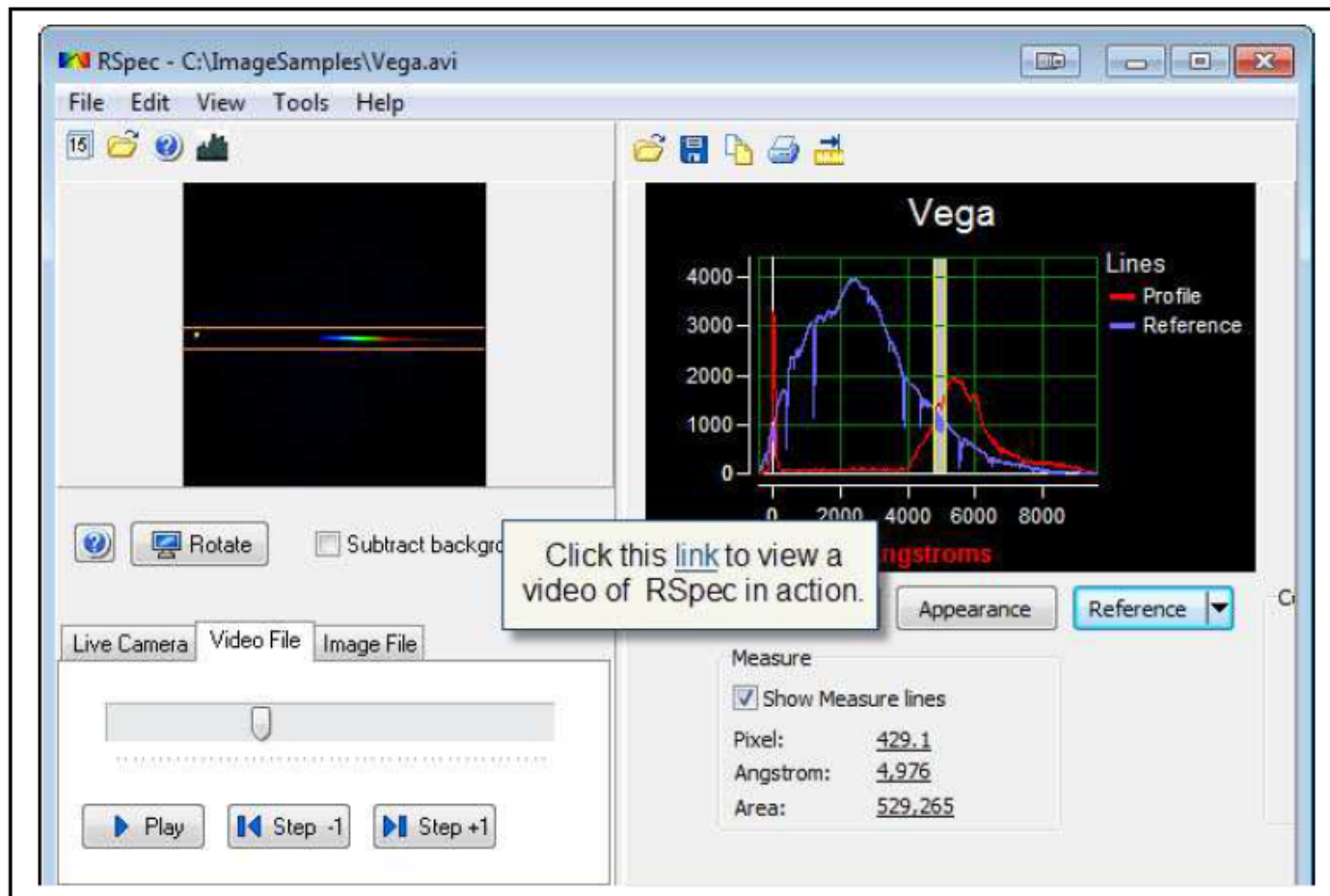
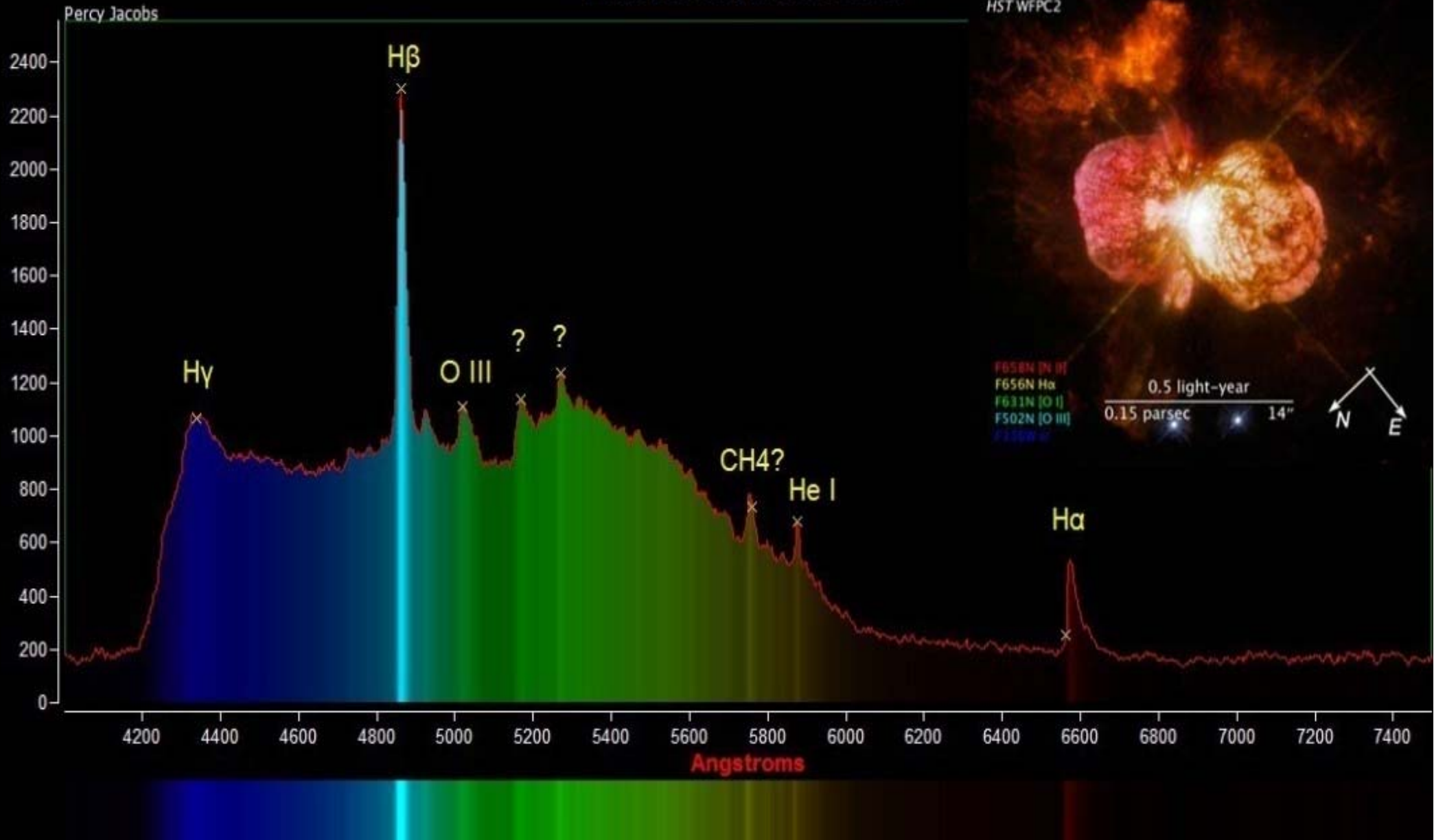


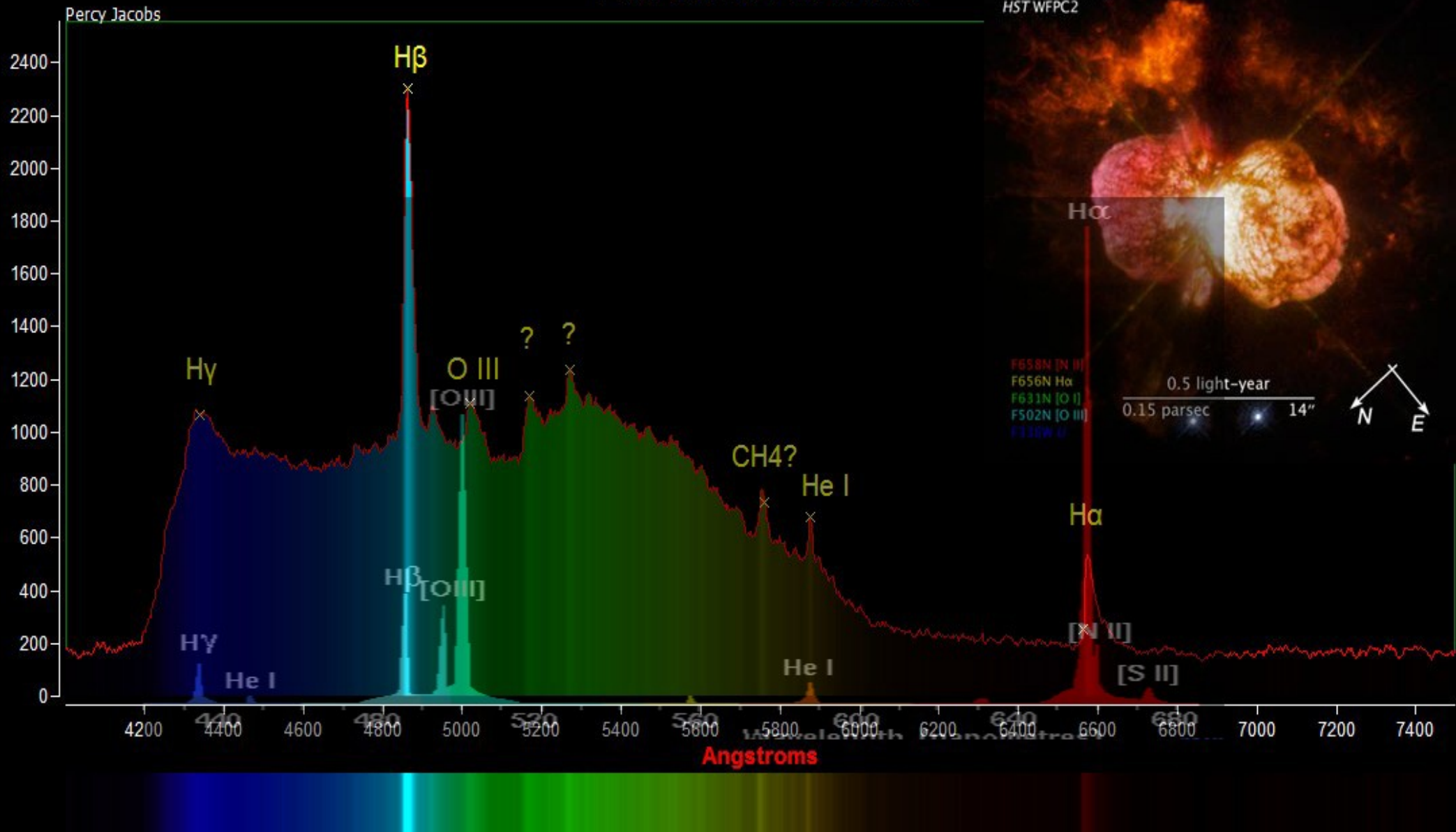
Fig. 6 Screenshot of the software used, RSPEC.

The RSPEC programme is comparing pixels to angstroms [2]. The star itself, the 1st calibration point, is taken as zero angstroms vs the pixel reading

number shown by the star. The 2nd calibration point would be the



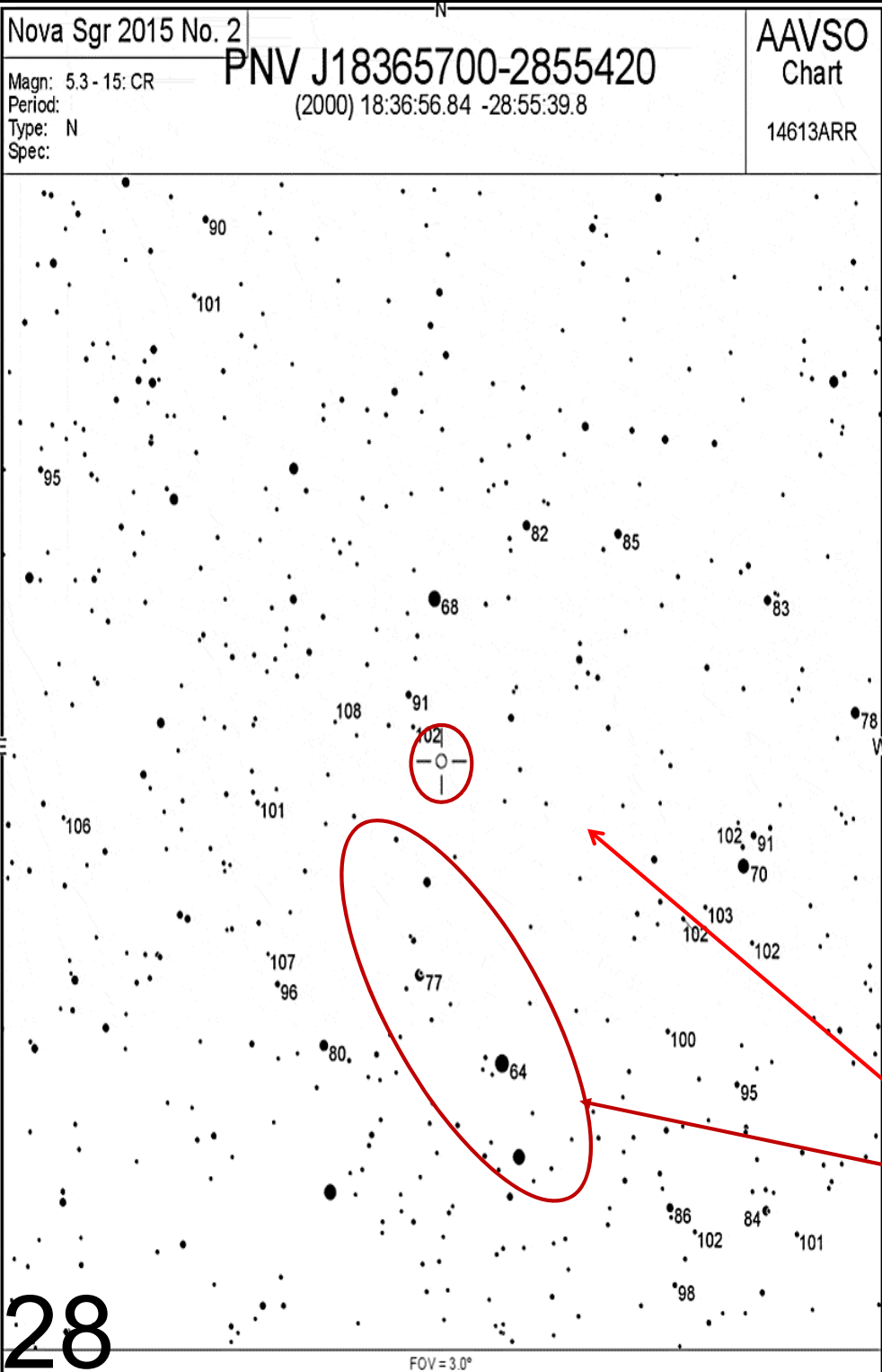
Showing the final processed image, is the spectrum of ETA Carina,
with the inserted picture of the actual star. The inserted picture is only
there to show what the target actually looks like.



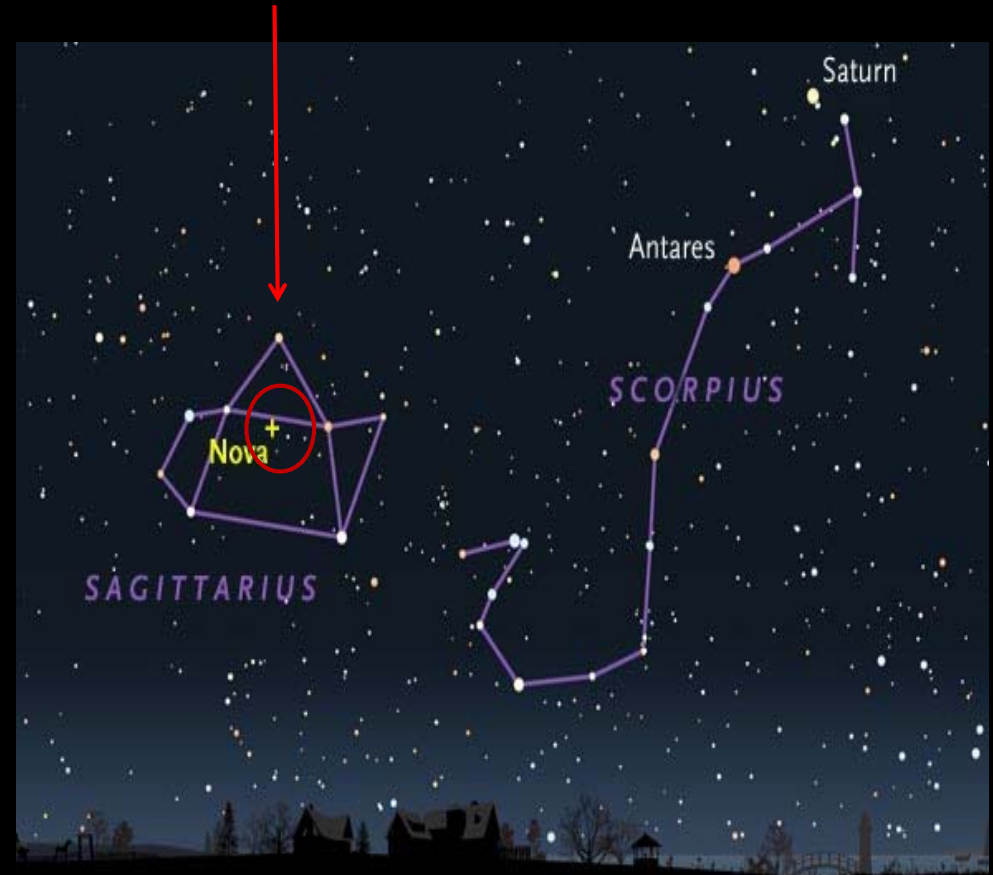
Doug Sharpe, located in the UK, superimposed a professional spectrograph on top of the one shown above and it matches pretty well.

NOVA SGR 2015 No.2

Spectra Analyses by Amateurs
using both, commercial & homemade equipment



Locating Nova Sgr 2015 No.2



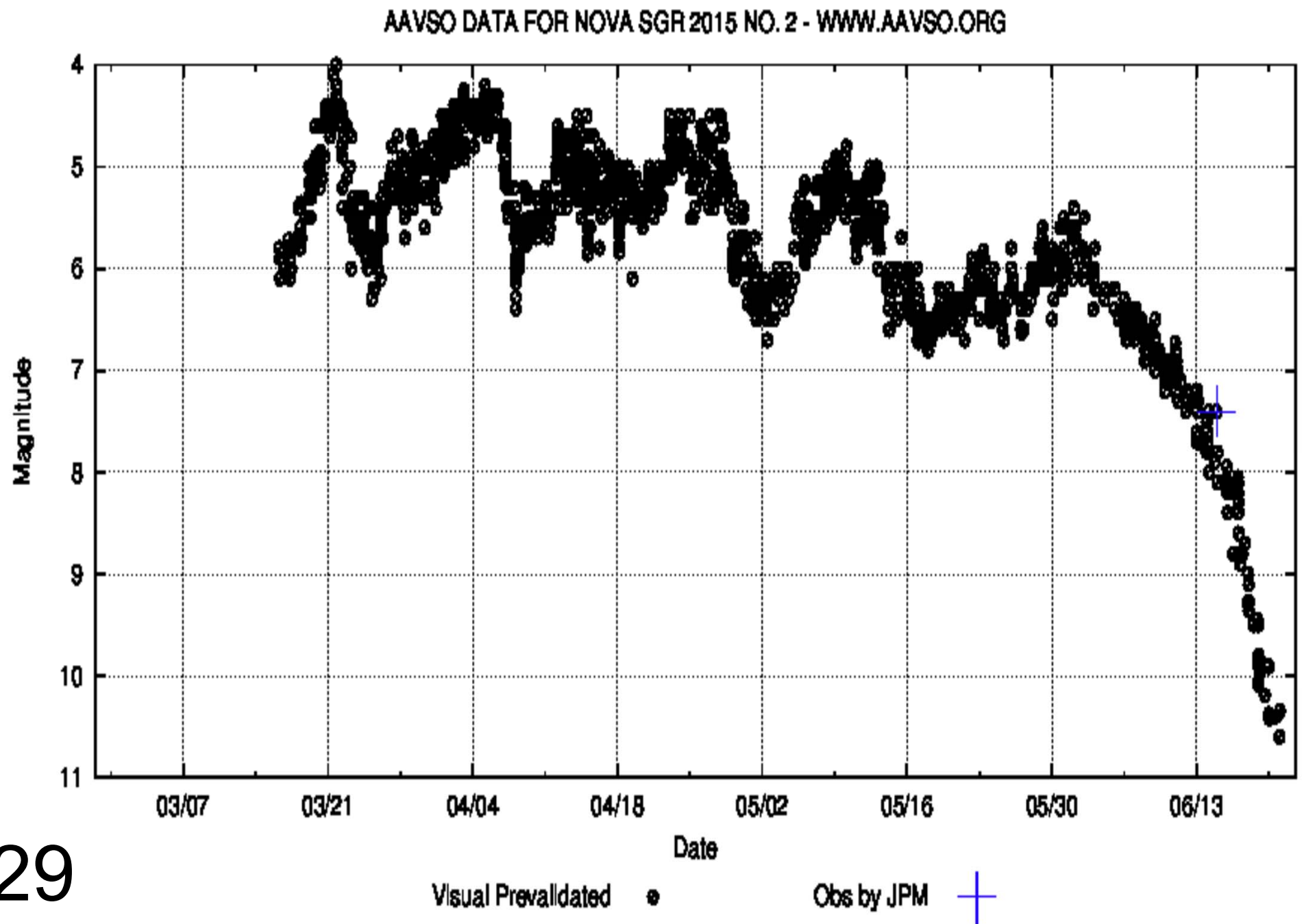
Looking South-Southeast

2nd step – hop to location using detailed AAVSO map

Star hop to location

Light Curve

visual magnitude - brightness



Spectra photo by Percy Jacobs of Pretoria ASSA

~7.5 magnitude - 14.06.2015

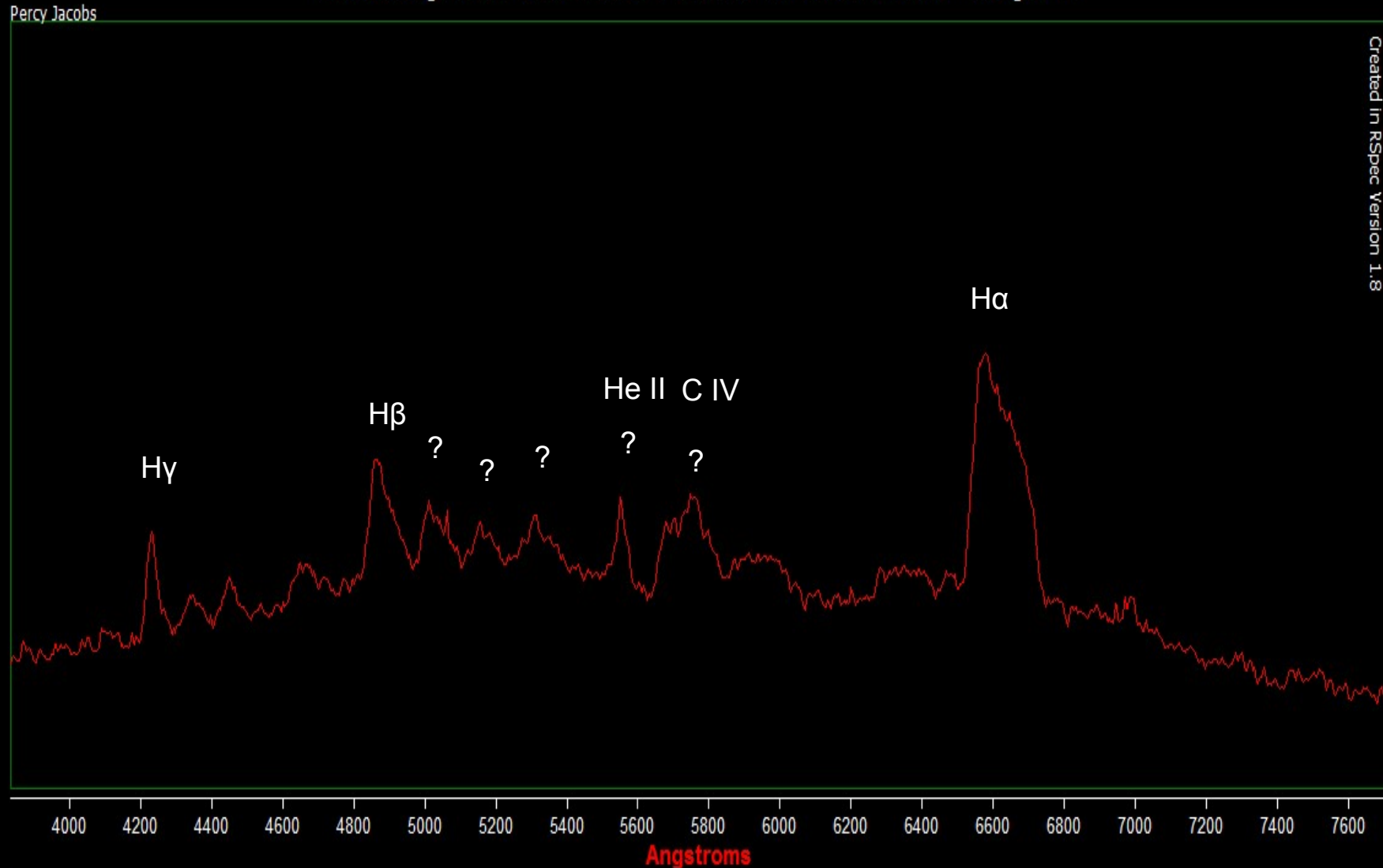


Spectra photo analysed using RSpec software

Percy Jacobs of Pretoria ASSA - ~7.5 magnitude

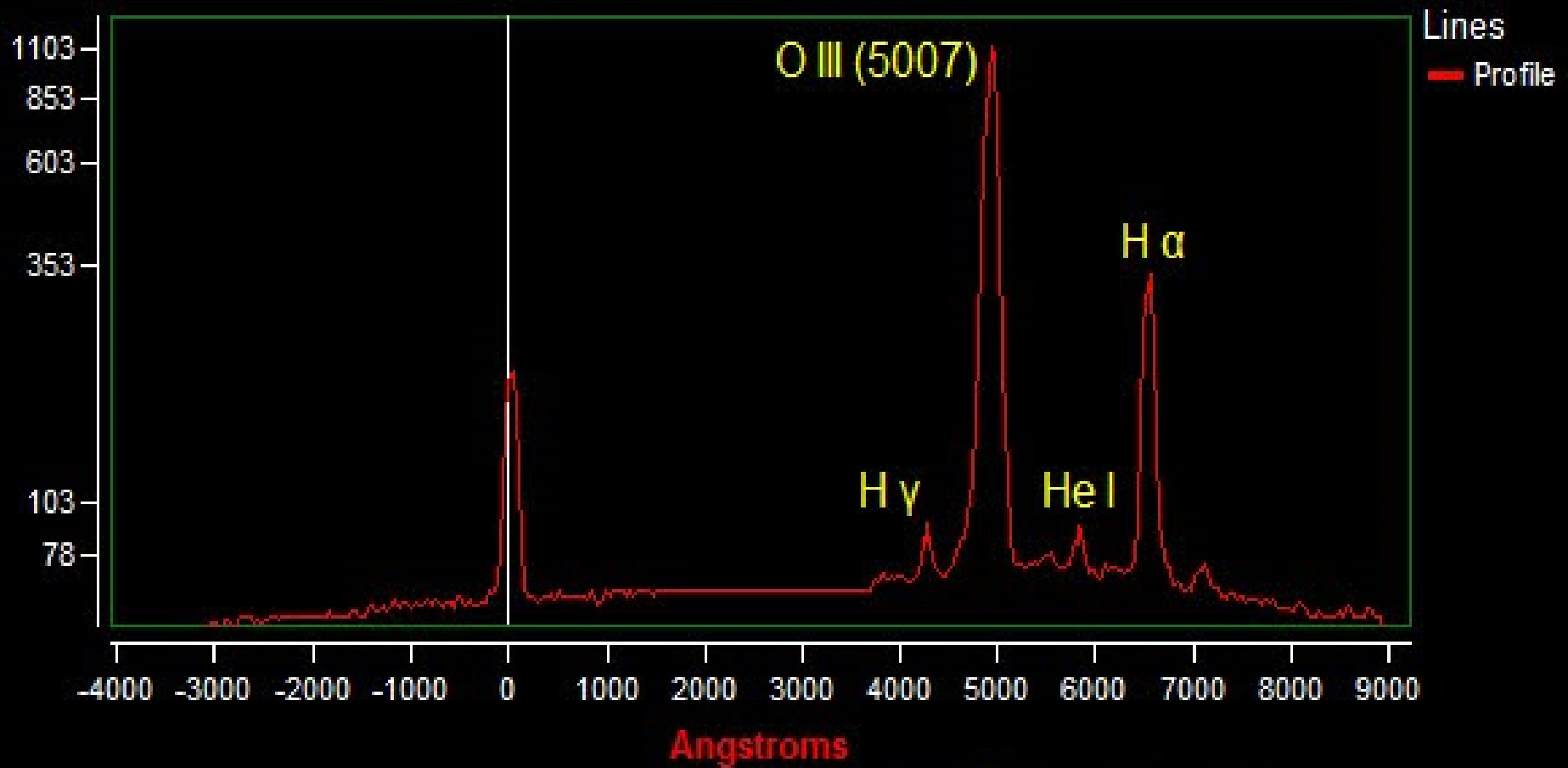
14.06.2015

Nova Sagittarii 2015 No. 2 14 June 15 South Africa ~mag. 7.4

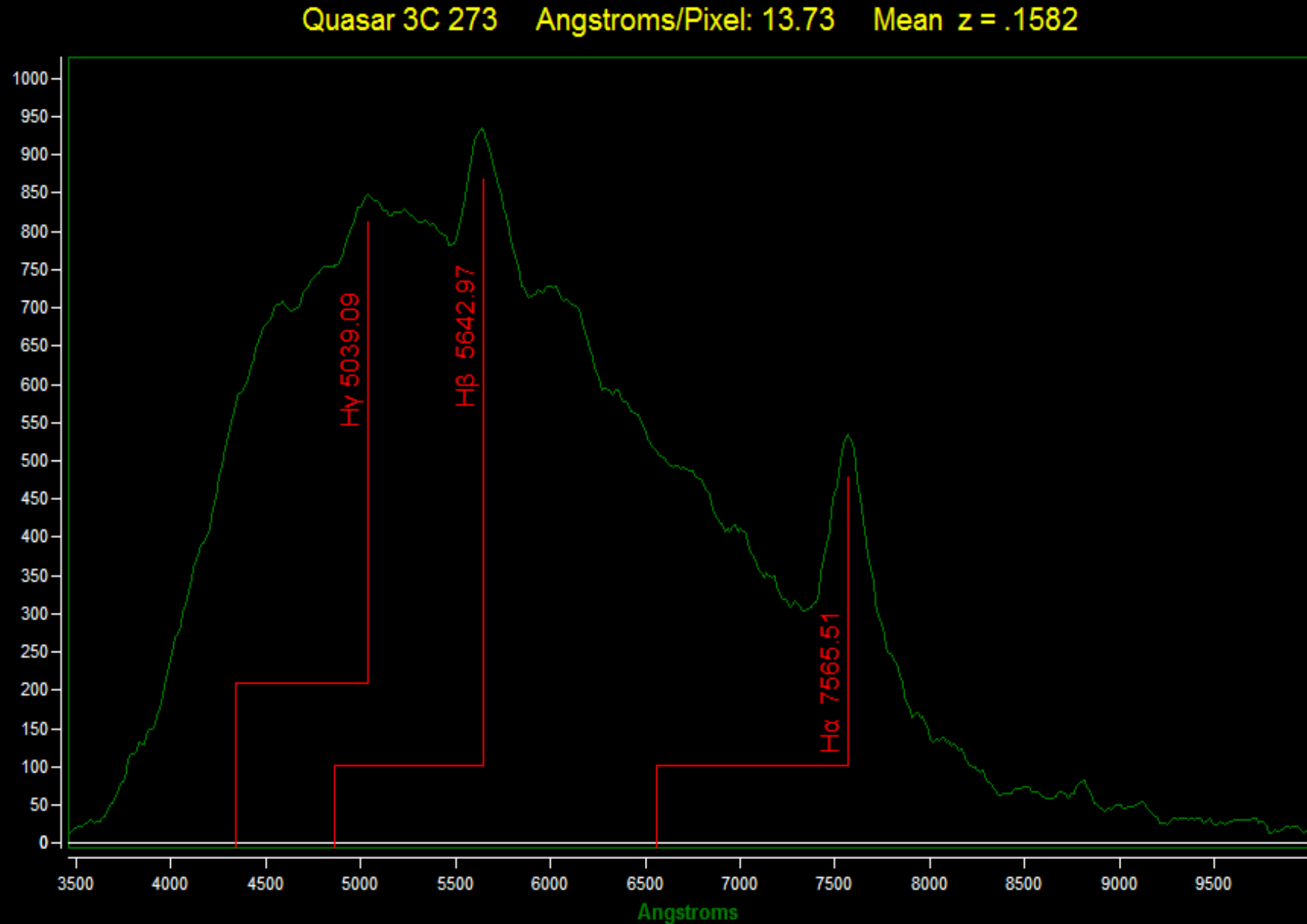


Detect the Emission lines on an Emission Nebula: An emission nebula is surrounded by a gas shell that is excited (like a fluorescent light). Below is a spectrum of the Saturn Nebula clearing showing its emission lines. The Orion Nebula (M42) would have a similar appearance.

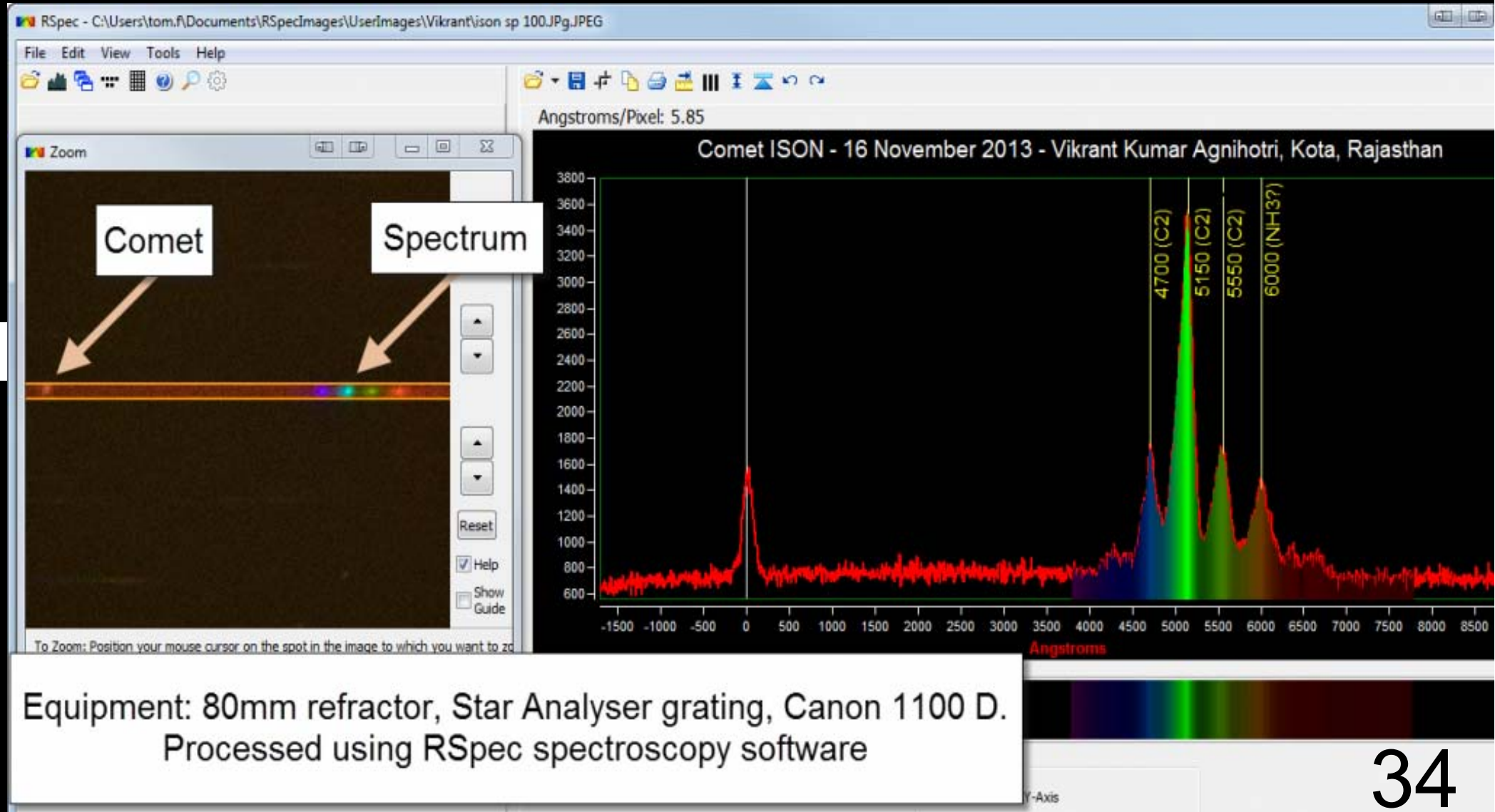
NGC 7009 - The Saturn Nebula



Detect the Red Shift of a Quasar that is 2 billion light years away. William Wiethoff's spectra (below) of QSO 3C 273 shows the red shift due to cosmological expansion. Many amateurs capture this spectrum on 8" telescopes with less than 15 minutes integration time.



Comet spectra: This spectrum of ISON shows how easy it is for amateurs to study astronomical spectra. Vikrant Kumar Agnihotri in India captured this wonderful spectrum of ISON using a just an 80 mm refractor, simple Star Analyser grating, and a DSLR. It clearly shows the green glow from glowing Carbon (the so-called “Swan bands” – Wikipedia [link](#)) This image was captured from a rooftop in Rajasthan, and then processed in the RSpec software:



Books

Spectroscopy: The Key to the Stars – Keith Robinson

Astronomical Spectroscopy for Amateurs – Ken M. Harrison

Astronomical Spectrography for Amateurs – EAS Publication Series – J.P. Rozelot, C. Neiner

Spectroscopic Atlas for Amateur Astronomers (no longer a free pdf download – now buy on-line through Cambridge University Press) -

Version 5.0 04/2014 (if you send me an email, I can send you a pdf copy of Version 4)

Software

Tom Field - RSpec – <http://www.rspec-astro.com> (most preferred – software comes with video tutorials that can be down loaded)

BASS Project (Basic Astronomical Spectroscopy Software by John Paraskeva – 2nd choice -

http://www.aesesas.com/mediapool/142/1423849/data/DOCUMENTOS/BASS_Project_1_.pdf

Visual Spec – <http://www.astrosurf.com/vdesnoux>

Christian Buil - <http://www.astrosurf.com/~buil>

Gratings

Rainbow Optics: manufacturer of the Star Spectroscope 200 l/mm grating - <http://www.starspectroscope.com/>

Paton Hawksley Star Analyser 100 (SA-100) - <http://www.rspec-astro.com/star-analyser/>

Groups

Astronomical Spectroscopy for Amateurs

https://groups.yahoo.com/neo/groups/RSpec_Real_Time_Spectroscopy/info

Basic Astronomical Spectroscopy Software

<https://uk.groups.yahoo.com/neo/groups/astrobodger/info>

Ken M. Harrison – very specialised in amateur spectroscopy and willing to help – ex member of the Durban ASSA group – you can contact him via the above “yahoo” group or direct on kenm.Harrison@gmail.com

Robin Leadbeater - THREE HILLS OBSERVATORY - (Formerly "ROBIN'S ASTRONOMY PAGE")

<http://www.threehillsobservatory.co.uk/astro/astro.htm>

A Good List of Links

<http://www.stargazing.net/david/spectroscopy/links.html>

Astronomical Society of Southern Africa

<http://assa.saao.ac.za/sections/photometry-spectroscopy/spectroscopy/>

Thank You